A study on Zn/C based Pathor Kuchi Leaf (PKL) electrochemical cell

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Abstract

It has been studied the performances of Zn/Cu based electrochemical cell previously. This paper shows the performances of Zn/C based PKL electrochemical cell. Carbon (C) electrode has been used instead of Copper (Cu) electrode. Different electric parameters like open circuit voltage, Short circuit current, maximum power, Load power, Load voltage, Load Current, Internal resistance have been studied. It is shown that the internal resistance lies between 3 ohm and 4.2 ohm, which is relatively high to the real values. Carbon (C) electrode use is the novelty of this research article.

I. Introduction

Carbon(C) is relatively low price with respect to the Cu electrodes. In this case no secondary salt has been used. PKL is available for everywhere. Because it grows everywhere. It has a great medicinal value. That is why it called miracle leaf. The zinc material is called anode material. It is also called sacrificial electrode. The both carbon and zinc materials are available in the local market and comparatively low price. This work may be the new guide line for near future. Oil, gas and coal are the mineral sources. It will be finished within short time all over the world. Then the renewable energy sources have to used. The renewable energy sources are solar energy, wind energy, geothermal energy, biogas energy, biomass energy, water energy, tidal energy, wave energy and OTEC etc. Our research work is the example of biomass energy.

Key words: Zn/C electrodes, PKL extract, PKL electrochemical cell, PKL electricity, Internal resistance

II. Methodology:

The total observation was for 90 hrs. After certain period the reading was collected by a calibrated multimeter.

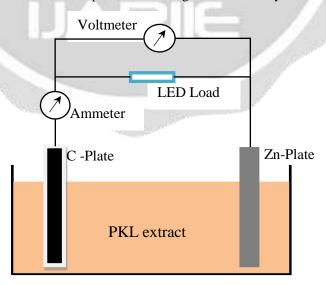


Fig.1 Basic principle of Zn/C based PKL electrochemical cell

Fig.1 shows the experimental set-up of the basic principle of Zn/C based PKL electrochemical cell. Here C-plate acts as a cathode and Zn-plate acts as an anode. PKL (Pathor Kuchi Leaf) acts as an electrolyte.



Fig.2 A PKL tree for leaf collection

Fig.2 shows the PKL tree. Leaves have been collected from this tree. After collection it has been blended by a blender machine. The produced extract has been filtered by the what man paper 41 and 42. The longevity of the PKL extract is around 12 years.



Fig.3 An experimental set up of Zn/C based electrochemical cell

Fig.3 shows an experimental set-up of a PKL electrochemical cell. There are 4- pairs of Zinc electrodes and 4 – pairs of Carbon electrodes. They are in series connection with the connecting wire. A calibrated multimeter has been used to measure the current, voltage and resistance.

III. Results and discussion

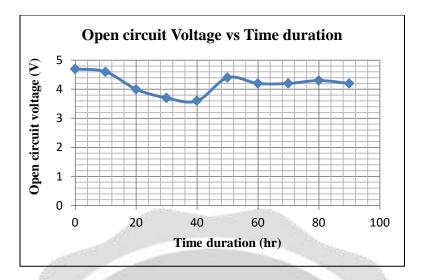


Fig.4 Open circuit Voltage vs Time duration

Fig.4 shows the Open circuit Voltage vs time duration curve. It is shown that the open circuit Voltage decreases slowly up to 40 hrs and then it increases exponentially up to 60 hrs and then after it was almost constant up to 90 hours.

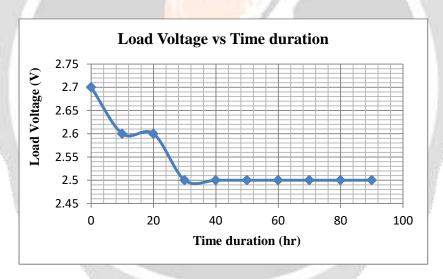


Fig.5 Load Voltage vs Time duration

Fig.5 shows the load voltage vs time duration curve. It is shown that the open circuit voltage decreases exponentially up to 20 hrs and then it decreases exponentially up to 40 hrs and then after finally it was almost constant up to 90 hours.

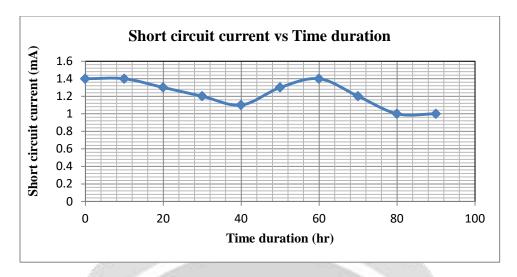


Fig.6 Short circuit current vs Time duration

Fig.6 shows the short circuit current vs time duration curve. It is shown that the short circuit current decreases up to 40 hrs and then it increase exponentially up to 80 hrs and then after finally it was almost constant up to 90 hours.

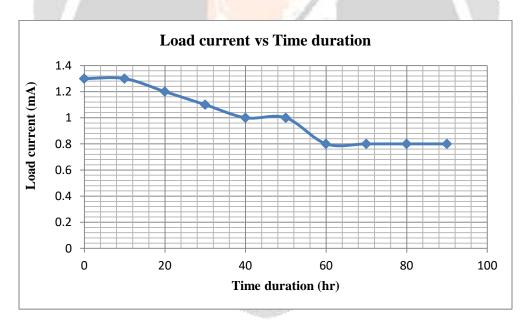


Fig.7 Load current vs Time duration

Fig.7 shows the load current vs time duration curve. It is shown that the load current decreases up to 40 hrs and then it increase exponentially up to 60 hrs and then after finally it was almost constant up to 90 hours.

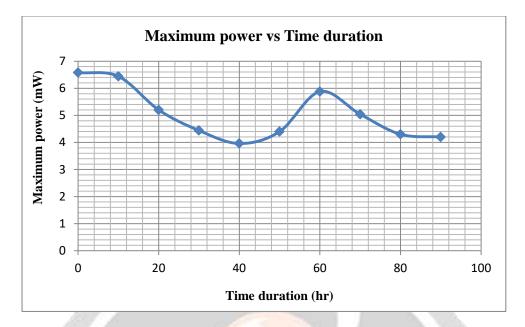


Fig.8 Maximum power vs Time duration

Fig.8 shows the maximum power vs time duration curve. It is shown that the maximum power decreases almost linearly up to 40 hrs and then it increase exponentially up to 80 hrs and then after finally it was almost constant up to 90 hours.

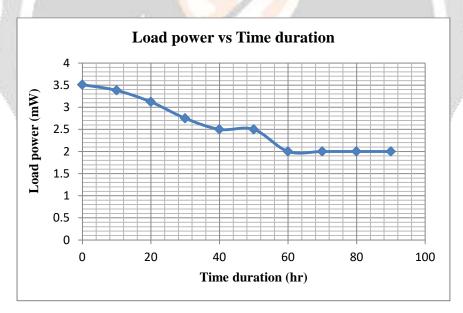


Fig.9 Load power vs Time duration

Fig.9 shows the load power vs time duration curve. It is shown that the load power decreases almost linearly up to 40 hrs and then it increase exponentially up to 60 hrs and then after finally it was almost constant up to 90 hours.

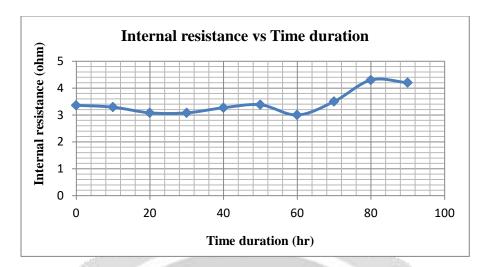


Fig.10 Internal resistance vs Time duration

Fig.10 shows the internal resistance vs time duration curve. It is shown that the internal resistances decreases almost linearly up to 30 hrs and then it increase exponentially up to 60 hrs and then after finally it increases almost exponentially up to 90 hours.

IV. Conclusions:

In this experiment no secondary salt was used. It needs to do further work using secondary salt. The internal resistance was relatively high. Next time internal resistance has to measure for secondary salt as comparative study.

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