

NEW METHOD DEVELOPMENT AND VALIDATION OF RP- HPLC FOR THE ESTIMATION OF IGURATIMOD IN THE TABLET DOSAGE FORM

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Abstract: The present study describes a simple and stability-indicating reverse phase high-performance liquid chromatography (RP-HPLC) method for the quantification of the related substances of Iguratimod drug substance. Successful separations of the possible impurities were achieved on a Inertsil ODS-3 (150 × 4.6 mm, 5µm) and UV detector at 257nm, 0.8mL/min as a flow rate, and 20µL as an injection volume. For mobile phase-A preparation, 5.44gm potassium dihydrogen orthophosphate dissolved in 2 liter water and adjusted pH 4.00±0.05 with dilute orthophosphoric acid and filter through 0.45µ membrane filter. Use HPLC grade Acetonitrile as mobile phase-B. Use gradient flow program with column temperature 40°C. Percentage recovery obtained in the range of 80-120% and the method is linear for all possible impurities and Iguratimod for specified concentration range with coefficient of variation (r) not less than 0.99. Acid, base, peroxide and thermal degradation were carried in drug substance. The proposed RP- HPLC method was found to be specific, linear, precise, accurate and robust.

Keywords: Iguratimod, validation, development, stability indicating methodology.

INTRODUCTION:

Chemically, Iguratimod 3-Formylamino-7-methylsulfonylamino-6-phenoxy-4H-1-benzopyran-4-one is used as an anti-inflammatory drug for the treatment of rheumatoid arthritis. It has following structure,

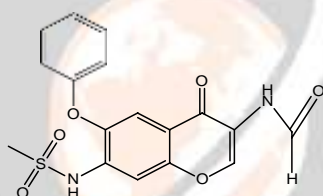


Fig 1: Structures of Iguratimod

IUPAC name is N-[(formylamino)-4-oxo-6-phenoxy-4Hchromen-7-yl] methane sulfoamide for Iguratimod. Iguratimod was first Reported in product patent US4954518.[1] Its Therapeutic category is Anti-arthritis and novel immunomodulator.[2] Iguratimod is a nuclear factor NF-κB activation inhibitor used in the treatment of rheumatoid arthritis. It also suppressed inflammatory cytokine production in cultured human synovial cells induced by tumor necrosis factor (TNF)-α by inhibiting the activity of nuclear factor-κB. Several synthesis processes are reported for Iguratimod. [3-6]. Efficacy of a drug substance is critical for its safety assessment. It is compulsory to identify and characterize the possible impurities in active drug. This compound is aromatic heterocyclic compound; belong to class of organic compound known as chromones. These are compounds containing a benzopyran-4-one moiety

Organic impurities in drug substances can arise during the manufacturing process and storage. Thus, the acceptance limits are based on pharmaceutical studies or known safety data. Several methods have been reported for the analysis of Iguratimod. However, no combined validated stability-indicating reversed phase HPLC (RP- HPLC) method has been used for the separation and quantitative analysis of all the possible impurities. In this study, a rapid and validated RP-HPLC method was developed to separate closely eluted impurities. The limit of detection (LOD), limit of quantification (LOQ) and sensitivity of the method was tested in accordance with ICH Q2 guidelines.

EXPERIMENTAL AND METHODS:

REAGENTS, MATERIALS AND INSTRUMENTATION

For analytical development activity, used HPLC grade methanol, acetonitrile and water. Analytical grade potassium dihydrogen orthophosphate and Orthophosphoric acid has been used. The instrument used was Waters HPLC system consisting of a pump, a UV detector and empower software. Analytical column with specification as Inertsil ODS-3 (150 × 4.6 mm, 5µm) was used for analysis.

OPTIMIZATION OF CHROMATOGRAPHIC CONDITIONS

Purpose of analytical method optimization is specifically to identify the analyte peak in presence possible impurities. The chromatographic separation of Iguratimod from its impurities was achieved using Inertsil ODS-3 (Dimension: Length 15 cm, 4.6 mm internal diameter and particle size 5µ) with mobile phase in gradient proportion at flow rate of 0.8 mL/min and detection wavelength of 257 nm.

Optimized chromatographic parameters and conditions	
Parameters	Chromatographic conditions
Stationary phase	Inertsil ODS-3, 150mm x 4.6mm, 5µ
Flow rate (Gradient)	0.8ml/min

Injection volume	20 μ l
Detection wavelength	UV 257 nm
Runtime	40.0 minutes
Column oven temperature	40°C
Diluent	Mobile phase-A : Mobile phase-B (50:50)

Table 1: Optimized chromatographic parameters and conditions

Gradient program :		
Time (minutes)	%Mobile phase-A	%Mobile phase-B
0	70	30
10	70	30
20	55	45
25	20	80
30	20	80
32	70	30
40	70	30

Table 2: Mobile phase preparation and gradient program

Preparation of mix standard solution

Prepare individually all stock solutions to get each 5ppm Impurity-A, Impurity-B, Impurity-D, Impurity-E , Igaratimod and 7.5ppm Impurity-C. Further dilute 2ml of each solution to 20ml with diluent to get 0.5 ppm Impurity-A, Impurity-B, Impurity-D, Impurity-E, Igaratimod, and 0.75 ppm Impurity-C.

Preparation of sensitivity solution

Dissolve about 5mg Igaratimod standard to 50ml with diluent, dilute 5ml to 50ml with diluent. Further dilute 1 ml of resultant solution to 100ml with diluent to get solution of 0.05ppm.

Preparation of test solution

Dissolve about 10mg sample to 20ml with diluent.

RESULTS AND DISCUSSION

OPTIMIZATION OF CHROMATOGRAPHIC CONDITIONS:

Many efforts have been made to develop a cost-effective, rapid, and robust reversed phase (RP)-HPLC method with enough data of validation parameters. First, pKa of drugs was investigated and pKa of Igaratimod was 2.96. As a rule of thumb, pH of mobile phase is selected two units above or below the pKa value of drug. If we consider pKa of Igaratimod, then we cannot choose the pH above 5.0, which is detrimental to silica beds of column. With respect to Igaratimod, we could choose the pH of mobile phase between pH 2.0 to 4.0. Therefore, we tried for pH about 3.5, which will be nearby pKa, and at this pH, Igaratimod will remain ionized, which makes better retention as well as separation of related compound. Thus, we tried at different pH and started at with pH 3.5, but at this pH all other impurities were separated and co-elution observed for unknown impurity at Acid impurity.

By changing the gradient ratio required separation achieved but principle peak shape deteriorated. By increasing the column oven temperature and reducing flow rate to 0.8ml/minutes, required separation of all the impurities achieved. During the pH parameter optimization, observed that at lower pH, co-elution of acid impurity with unknown impurity at higher pH, good separation achieved. During analytical column change, decrease in resolution between two known impurities were observed and got learning that decreasing the pH of mobile phase, resolution decreases. Hence slight change in pH on higher side, i.e. increase in pH, 3.9 to 4.0 was finalized for better separation for all possible known impurities and between peak shape.

By using the pH 4.0, it gives sharp peak of Igaratimod without co-elution or any interference of unknown or known impurities. Optimized chromatographic parameters are summarized in Table1 and Table 2. Typical chromatogram is shown in Fig. 2, Fig. 3, Fig. 4, and Fig. 5. This study was validated according to the guidelines of International Conference on Harmonization (ICH) and USP.

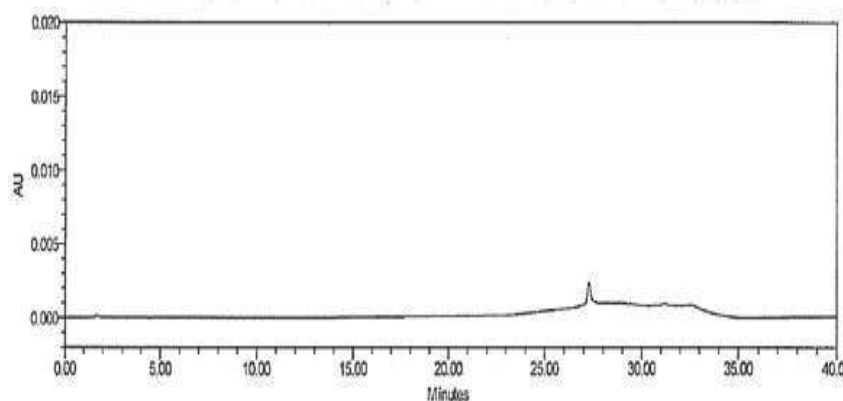


Fig 2: Blank chromatogram

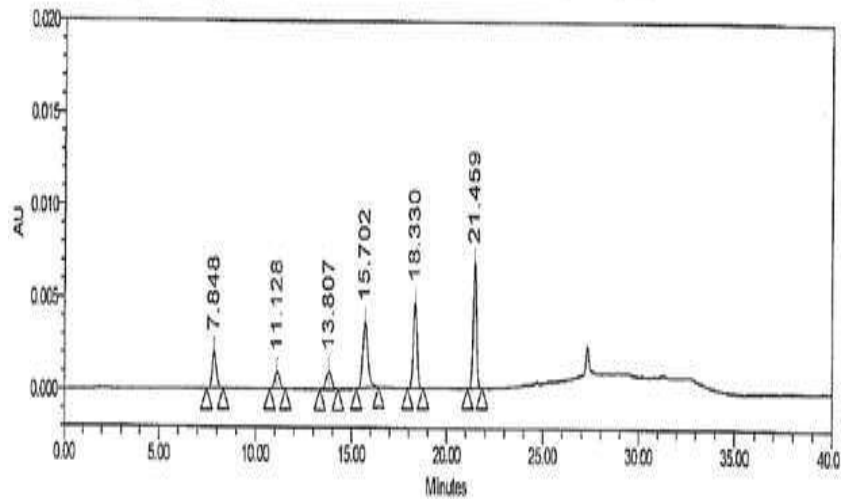


Fig 3: Mix standard chromatogram

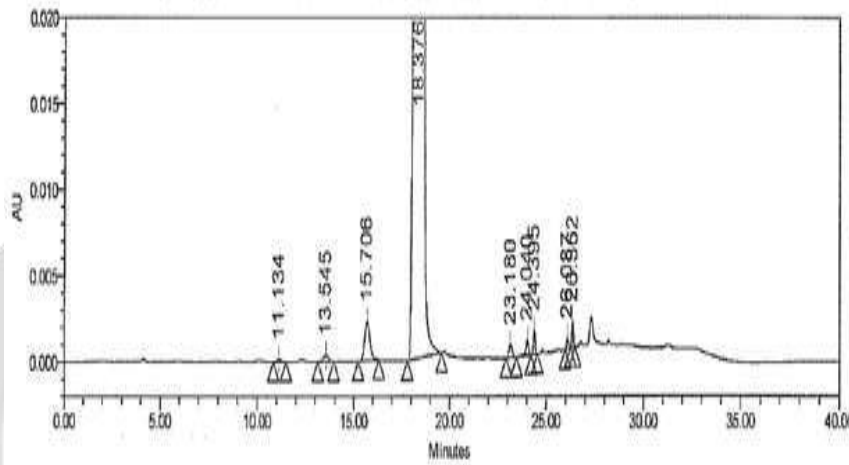


Fig 4: Test (as such) chromatogram

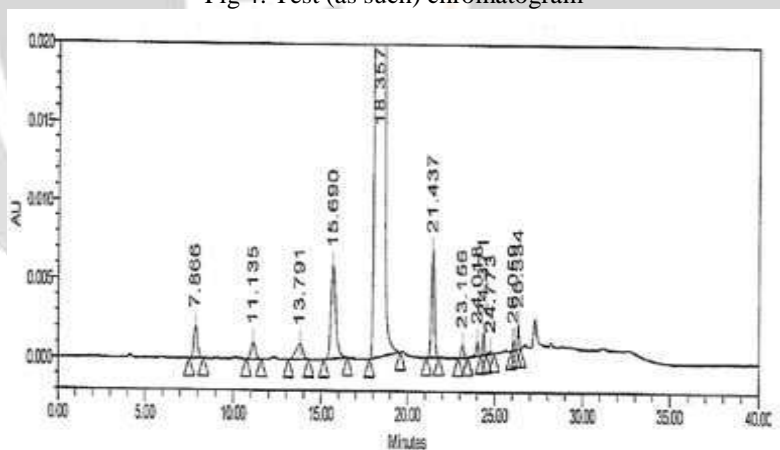


Fig 5: Test spiked with known impurities chromatogram

**METHOD VALIDATION:
SPECIFICITY:**

Blank, mix impurity standard, Test (as such), Test spiked with impurities were injected to evaluate Specificity. No interference due to blank and coelution of impurities were observed. All known peaks pass peak purity criteria i.e. purity angle should be less than purity threshold.

To demonstrate the stability indicating nature of the method, forced degradation has been carried out in acid (1N Hydrochloric acid, 60°C, 30 minutes), base (0.1N Sodium hydroxide, at room temperature for 5 minutes), oxidation (10 % Hydrogen peroxide at 60°C for 4 hours), aqueous degradation (Water at 60°C for 4 hours).

In acidic condition Impurity-E increases up to 5%, where as in alkaline condition unknown impurity increase upto 9.5%. No significant degradation observed in aqueous and oxidation degradation condition.

The entire peaks were found to be resolved form each other and spectrally pure as calculated by empower software.

Specificity					
Peak Name	Retention time (minutes)	Peak Tailing (USP)	Purity angle	Purity Threshold	Peak Purity
Impurity-D	7.484	2.440	0.298	1.709	Passes
Impurity-A	11.128	1.060	0.247	0.251	Passes
Impurity-B	13.807	1.040	0.206	0.252	Passes
Impurity-E	15.702	1.090	0.160	0.266	Passes
Iguratimod	18.330	1.000	0.230	0.265	Passes
Impurity-C	21.459	1.020	0.250	0.380	Passes

Table 3: Specificity

PRECISION

Consistency in repeated response by the chromatographic system and solution preparation procedure was evaluated by injecting six replicate preparations of mix standard spiked with impurities. % RSD for retention time and area response is tabulated in Table 4 and Table 5 respectively.

Component Summary For Retention Time
Channel: W2489 ChA

	Sample Name	Inj	Channel	Vial	Result Id	Acid Impurity	D-SPA.F	SPA.F	Acid FD	Iguratimod	Methyl Impurity
1	Mix Standard	1	W2489 ChA	51	4945	6.597	10.234	12.594	14.561	17.345	20.547
2	Mix Standard	2	W2489 ChA	51	4946	6.589	10.224	12.578	14.553	17.343	20.550
3	Mix Standard	3	W2489 ChA	51	4947	6.590	10.227	12.589	14.565	17.358	20.567
4	Mix Standard	4	W2489 ChA	51	4948	6.595	10.232	12.590	14.569	17.351	20.557
5	Mix Standard	5	W2489 ChA	51	4949	6.585	10.220	12.575	14.548	17.339	20.546
6	Mix Standard	6	W2489 ChA	51	4950	6.579	10.217	12.572	14.547	17.345	20.557
Mean						6.589	10.226	12.583	14.555	17.347	20.554
Std. Dev.						0.007	0.006	0.009	0.007	0.007	0.008
% RSD						0.1	0.1	0.1	0.0	0.0	0.0

Table 4: Precision for retention time

	Sample Name	Inj	Channel	Vial	Acid Impurity	D-SPA.F	SPA.F	Acid FD	Iguratimod	Methyl Impurity
1	Mix Standard	1	W2489 ChA	51	20821	15739	19466	47043	67468	88680
2	Mix Standard	2	W2489 ChA	51	20822	15741	19510	46825	67479	88581
3	Mix Standard	3	W2489 ChA	51	21005	15916	19539	47161	67847	88993
4	Mix Standard	4	W2489 ChA	51	20844	15684	19324	46630	67173	88189
5	Mix Standard	5	W2489 ChA	51	21110	15884	19713	47113	67900	89080
6	Mix Standard	6	W2489 ChA	51	20925	15865	19618	47100	67814	89171
Mean					20921	15805	19528	46979	67613	88782
Std. Dev.					117	95	133	208	286	371
% RSD					0.6	0.6	0.7	0.4	0.4	0.4

Table 5: Precision for area response

LINEARITY

To evaluate the linearity in detector response all the components were injected from LOD level to 150% wrt test concentration and the correlation coefficient was found to be not more than 0.99. The response factor was calculated from the slope of impurities and Iguratimod linearity curve; all impurities have RRF below 1.0.

Correlation coefficient

Conc. (in ppm) for Impurity-D	Avg Area
0.0763	2881
0.2289	9041
0.7631	29534
1.1446	46402
Slope =	40380.6979
Correlation coefficient =	0.9995
Squared Correlation coefficient =	0.9990
Relative Response Factor =	0.2962

Table 6: Linearity for Impurity-D

Conc. (in ppm) for Impurity-A	Avg Area
0.0534	1553
0.1601	4943
0.5338	16024
0.8007	25098
Slope =	31214.1747
Correlation coefficient =	0.9996
Squared Correlation coefficient =	0.9991
Relative Response Factor =	0.2290

Table 7: Linearity for Impurity-A

Conc. (in ppm) for Impurity-B	Avg Area
0.0509	1842
0.1526	5917
0.5085	19265
0.7628	30182
Slope =	39447.3062
Correlation coefficient =	0.9996
Squared Correlation coefficient =	0.9991
Relative Response Factor =	0.2893

Table 8: Linearity for Impurity-B

Conc. (in ppm) for Impurity-E	Avg Area
0.0812	6506
0.2345	21167
0.8118	70877
1.2177	112980
Slope =	92449.4200
Correlation coefficient =	0.9992
Squared Correlation coefficient =	0.9983
Relative Response Factor =	0.6781

Table 9: Linearity for Impurity-E

Conc. (in ppm) for Iguratimod	Avg Area
0.0505	6521
0.1514	20304
0.5047	65353
0.7571	103967
Slope =	136335.3554
Correlation coefficient =	0.9992
Squared Correlation coefficient =	0.9983
Relative Response Factor =	1.0000

Table 10: Linearity for Iguratimod

Conc. (in ppm) for Impurity-C	Avg Area
0.0785	8260
0.2354	25793
0.7848	83303
1.1772	130353
Slope =	110115.0903
Correlation coefficient =	0.9996
Squared Correlation coefficient =	0.9991
Relative Response Factor =	0.8077

Table 11: Linearity for Impurity-C

LIMIT OF DETECTION AND LIMIT OF QUANTIFICATION:

The Limit of detection (LOD) and limit of quantification (LOQ) were determined on visual basis. The LOD and LOQ values are depicted in Table 12.

LOD & LOQ Determination		
Peak Name	LOD (wrt test conc.)	LOQ (wrt test conc.)
Impurity-D	0.015	0.045
Impurity-A	0.010	0.030
Impurity-B	0.010	0.030
Impurity-E	0.015	0.045
Impurity-C	0.015	0.045

Table 12: LOD and LOQ determination

ACCURACY:

Accuracy was demonstrated by spiking impurity solution at LOQ, 100 % and 150 % of test concentration. % Recovery was calculated

from the amount added and amount found. The results are tabulated in Table 13.

Accuracy				
Peak name	%Recovery at 30% level	%Recovery at 80% level	%Recovery at 100% level	%Recovery at 150% level
Impurity-A	95.455	98.555	96.903	95.700
Impurity-B	88.730	97.435	96.246	95.270
Impurity-C	94.915	97.335	96.000	95.130
Impurity-D	99.350	99.945	98.337	97.485
Impurity-E	95.675	101.045	99.300	98.030

Table 13: Accuracy

ROBUSTNESS:

Robustness of the method was evaluated by injecting test spiked with impurities, retention time and area response was monitored. Analytical method was found to be robust for the below mentioned conditions,

1. ± 0.1 mL flow rate
2. ± 0.2 unit's pH of buffer
3. ± 5 °C column temperature

Robustness (Change in pH by ± 0.20)				
Peak name	pH - 0.20		pH + 0.20	
	Retention Time (minutes)	Area response	Retention Time (minutes)	Area response
Impurity-A	10.784	16050	10.778	16339
Impurity-B	13.352	19496	13.363	19770
Impurity-C	21.049	85193	21.059	87064
Impurity-D	7.788	31381	6.776	31538
Impurity-E	15.269	66871	15.282	69400

Table 14: Robustness: Change in pH

Robustness (Change in column temperature $\pm 5^\circ\text{C}$)				
Peak name	Column temperature -5°C		Column temperature $+5^\circ\text{C}$	
	Retention Time (minutes)	Area response	Retention Time (minutes)	Area response
Impurity-A	11.115	17462	9.519	17577
Impurity-B	13.567	21301	11.747	21359
Impurity-C	21.338	90982	19.865	90988
Impurity-D	7.968	31514	6.691	32081
Impurity-E	15.576	72189	13.687	72539

Table 15: Robustness: Change in column temperature

Robustness (Change in flow rate by ± 0.1 ml/minute)				
Peak name	Flow - 0.1 ml/min		Flow + 0.1 ml/min	
	Retention Time (minutes)	Area response	Retention Time (minutes)	Area response
Impurity-A	11.634	19537	9.174	15438
Impurity-B	14.269	23587	11.273	18674
Impurity-C	21.919	101635	19.478	79533
Impurity-D	8.213	34876	60484.000	27567
Impurity-E	16.224	81184	13.221	62463

Table 16: Robustness: Change in column flow rate

CONCLUSION:

The RP-HPLC method developed for the estimation of Impurities in Igaratimod. This analytical method was found to be specific, precise, robust, linear and accurate. The method can be used for checking the quality of Igaratimod for routine as well as stability studies.

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